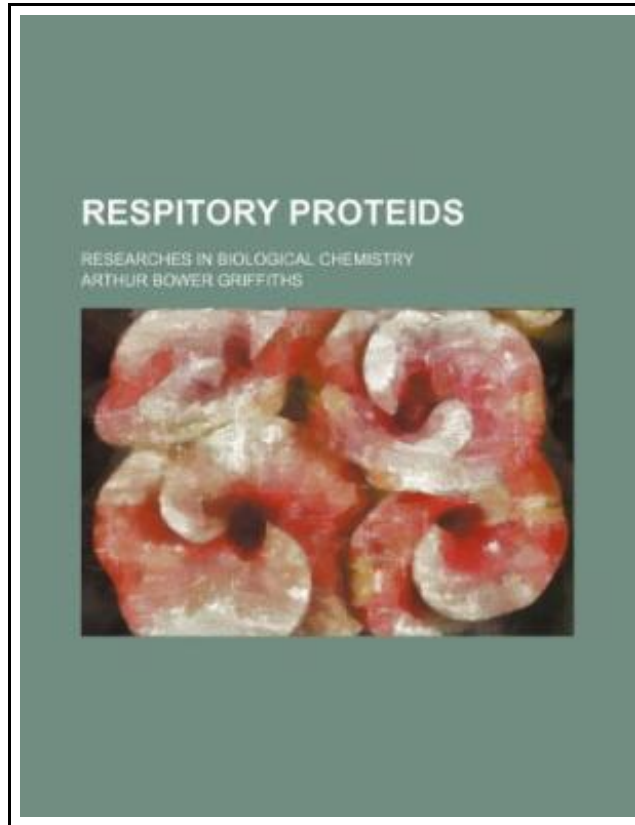


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1897 Excerpt: .needles. It is soluble in boiling absolute alcohol; but is more soluble in ether, carbon disulphide, benzene, and aniline than in alcohol. Its best solvent is chloroform. Its solutions are less distinctly green and more brown than those of phyllocyanin; they are fluorescent, and when dilute exhibit a reddish tinge, of which nothing is seen in the case of phyllocyanin. The spectra of phylloxanthin and phyllocyanin are given in Fig. 3. Phylloxanthin, like phyllocyanin, may be heated for several hours at i30C. without undergoing any change, but at i60C. decomposition commences and at 180C. the substance is completely decomposed. When incinerated in a platinum dish, phylloxanthin leaves an ash which consists of ferric oxide. Schunck has proved that phylloxanthin always leaves an ash of ferric oxide after incineration. This appears to prove that iron in some form or other is an essential constituent of chlorophyll or leaf-green, and that the strong acids used in the prepara Spectra Of Chlorophyll, Etc. i. Chlorophyll (in alcohol). 2. Phyllocyanin (in ether). 3. Phylloxanthin (in ether). 4. Alkachlorophyll (in ether). 5. Phylloporphyrin (in ether). 6. Hfematoporphyrin (in ether). tion of phylloxanthin do not interfere with the iron which is in organic combination. When cupric acetate is added to a solution of phylloxanthin a compound is formed which is similar to phyllocyanin cupric acetate. It crystallises in small scales, which are purple and lustrous by reflected light and pale green by transmitted light. No phylloxanthin zinc acetate has been obtained; but...



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